Research Article

Anticancer activity of novel surface functionalized metal doped hydroxyapatite

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Abstract

In this manuscript it has been described a novel synthesis of mercury doped hydroxyapatite (Hap) and its application on human liver carcinoma cell line (Hep G2) and human lung fibroblast carcinoma cell line (MRC 5). Nano-sized hydroxyapatite doped with Hg²⁺ was synthesized by a solution based chemical method along with mercury ion. The surface of nanoparticle of mercury doped hydroxyapatite (MHAp) was functionalized by using phosphonomethyl iminodiacetic acid (PMIDA) for making it stable as dispersed phase with negative zeta potential. Surface functionalization was confirmed by FTIR measurements. Crystalline nature, morphology and surface topology were studied by powder XRD, FESEM and AFM measurements. Particle size of the well dispersed sample was obtained by HRTEM image. The studies on cell viability of Hep G2 and MRC 5 cell in presence of mercury doped HAp nanoparticle (MHAp) were determined through WST assay. It was observed that nanocomposite exhibited a site specific action towards MRC 5 cell lines along with reduction of toxicity toward normal cells.

Introduction

Nanotechnology represents a new and enabling platform that promises to provide a range of innovative protocol for biomedical applications. Nanosized materials, such as nanoparticles, nanorods, nanowires, nanotubes and nanofiber have been explored in many biomedical applications because of their novel properties and multi functionality. Hydroxyapatite $[Ca_{10}(PO_4)_6(OH)_2]$ (HAP) is an interesting biomaterial with potential orthopedic and dental applications due to its excellent biocompatibility, bioactivity and osteoconductivity [1-3]. It is however worth-mentioning that the application of pure HAP is limited due to its hardness and chemical stability. However in recent years HAp-polymer composite has gained importance with good biological and mechanical performances [4-6] along with dispersion in collagen matrix [7]. Sometimes nano-hydroxyapatite (nHAP) crystals are dispersed in suitable polymeric matrices [8–12] along with organophosphate and phosphonic acid based coupling agents [13-20] for various applications.

Transition metals are widely known for their in-vitro toxicity; however considerable efforts are being made towards their application in anticancer activity with their site specific targeting ability [21-23] to reduce infectious **More Information**

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Submitted: 25 May 2020 Approved: 01 June 2020 Published: 03 June 2020

How to cite this article: Pradhan S, Mitra S, Goswami A. Anticancer activity of novel surface functionalized metal doped hydroxyapatite. Arch Pharm Pharma Sci. 2020; 4: 021-026.

DOI: 10.29328/journal.apps.1001023

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Keywords: Hydroxyapatite; Nanocomposite; Surface grafting; Cytotoxicity; Cell viability





cancer cells. Mercury and its compounds are widely known for its high toxicities. Constant consumption of mercury (II) causes harmful disease like Minamata, Acrodynia and Hunter-Russell syndrome. However very low concentration of mercury delivery to cancer affected cell lines may act as alternative anticancer agent whose toxicity is expected to be less than that of Pt-compounds used as anti-cancer agent. Hydroxyapatite is bio-compatible and nontoxic in nature, therefore doping of mercury metal ion in low concentration will reduce the toxicity of metal ion to some extent. Mercury based compounds such as phenyl mercuric nitrate and phenyl mercuric acetate had been used as antimicrobial agents especially phenyl mercuric nitrate [24,25] however no research reports are there of using mercury as anti-cancer agent. Oral administration of mercury based compounds causes toxicity due to their nonspecific binding towards proteins therefore our aim is to synthesize mercury doped nanocomposite into a biocompatible nanomaterial in order to remove nonspecific binding with proteins and avoid unexpected toxicity.

In our study a novel methodology has been developed to embed the mercury ion into the nanoparticle of hydroxyapatite via sol-gel technique. In order to avoid the phagocytocis of

the nanocomposite; the surface was functionalized by using phosphonomethyl iminodiacetic acid. Therefore grafting of carboxylic acid groups on to the surface of the particles render the nanocomposite to be more bio-compatible and stable dispersion in water or biological fluid. Mercury doped surface functionalized hydroxyapatite was then subjected to liver carcinoma cell line (Hep G2) and lung carcinoma fibroblast cell line (MRC 5) for in vitro cytotoxicological study. It is interesting to note that nanocomposite (MHAp) showed toxicity to the MRC 5 cell line whereas no toxicity was revealed towards Hep G2 cell line. Cytotoxicological study with mercury doped hydroxyapatite (MHAp) has not been reported till date to best of our knowledge. Therefore we were motivated to develop an unique synthesis of mercury doped surface functionalized nanoparticle of hydroxyapatite (nMHAP) and their applications as anti-cancer agents. The material has two major advantages:

(1) The precursors used in the whole process are inexpensive and easy to synthesize.

(2) Nanoparticle of MHAp has site specific anticancer activity towards lung carcinoma fibroblast cell line (MRC 5) whereas no toxicity was observed towards liver carcinoma cell line (Hep G2). Therefore reduction of toxicity in the nanocomposite switch the nanocomposite as anticancer agent towards lung fibroblast cell line.

Materials and methods

Chemicals

Calcium nitrate (99%), diammonium hydrogen phosphate (DAHP) (99%), mercuric chloride were obtained from E. Merck (India). Triethanolamine (TEA) (98%) and phosphonomethyl iminodiacetic acid were purchased from Sigma-Aldrich. Ammonia solution (25%), ammonium chloride (98%), and ethanol were also obtained from E. Merck (India). All the chemicals were of analytical grades and used without further purification. Milli-Q grade water (Sartorious Stedim Biotech) was used for all bio-assay with its conductivity less than 0.1 μ S cm⁻¹.

Cell lines

MRC 5 and HepG2 cell lines were purchased from ATCC (catalog no. CCL-171and HB-8065). All cultures were maintained in a phenol red free culture medium DMEM/ F12 (Dulbecco's modified essential medium/Ham's 12 nutrient mixture, Gibco), supplemented with 5% (v/v) fetal calf serum (JS Bioscience, Australia), and 1% (v/v) antibiotic (2 mM L-glutamine, 100 U/mL Penicillin and 0.1 mg/mL Streptomycin; Gibco). Cultured cells were kept at 37°C in a humidified 5% CO_2 incubator. Once the cells reached confluence, the culture medium was removed from the flask and the cells were rinsed three times with sterile HBSS (Hank's Balanced Salt Solution, Gibco). The confluent

cell layers were enzymatically removed, using Trypsin/ EDTA (Gibco, USA), and re-suspended in culture medium. Cell viability was assessed by vital staining with trypan blue (0.4% (w/v); Sigma, USA), and cell number was determined using an optical microscope (Carl Zeiss).

Experimental section

Synthesis of mercury doped nano hydroxyapatite

Mercury doped hydroxyapatite was synthesized by modification of a previously reported method [14]. In brief, stock solution of 0.5 (M) of calcium nitrate $Ca(NO_3)_2.4H_2O_1$ 0.5 (M) of diammonium hydrogen phosphate (NH4)₂HPO₄ (DAHP) and 0.01 (M) of mercuric chloride solution was prepared in distilled water. Calcium nitrate and diammonium hydrogen phosphate solution was taken in such an amount that the net atomic ratio of Ca:P was maintained at 1.67. Triethanolamine (TEA) was used in conjugation with calcium nitrate as the size controlling agent during its precipitation. The final pH of the reaction mixture of calcium nitrate and diammonium hydrogen phosphate solution was maintained throughout at ~11-12 using dilute ammonia. DAHP solution and 10 ml of 0.01 (M) mercuric chloride solutions was added drop wise simultaneously to the mixture of calcium nitrate and triethanolamine and stirred vigorously at ambient temperature with the help of a mechanical stirrer with 3000 rpm. The reaction mixture was allowed to stir for about 6 hours and produced a gelatinous precipitate. The precipitate was separated by centrifugation at 4000 rpm for 15 minutes. The precipitate was washed thoroughly with water and ethanolic ammonium chloride to remove excess of TEA. Finally the obtained product was dried at 90°C for 15 hours.

Grafting of carboxylic acid on the surface

functionalized Carboxylic acid mercury doped hydroxyapatite (nMHAP) was synthesized by using phosphonomethyl iminodiacetic acid (PMIDA). In brief, about 500 mg of the nanocomposite was taken in 25 ml of distilled water and kept in a sonication bath for about 45 minutes until a well dispersed solution was obtained. To this well dispersed sample solution aqueous alkaline solution of PMIDA was added drop wise maintaining the molar proportion of 1:1, alkaline pH 9 was maintained throughout the addition process. The reaction mixture was allowed to stir for 6 hours; finally the product was separated by centrifugation at 4000 rpm for 15 minutes. The product was washed with distilled water for several times to remove excess of PMIDA and dried at a temperature of 60 °C. Carboxylic acid grafted nanocomposite was very stable in aqueous dispersion; this functionalized composite was used for bio-assay.

Characterization

Morphology of the nanocomposite was investigated by using SUPRA 40 Field Emission Scanning Electron Microscope

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of Carl Zeiss with acceleration voltage of 5.0 kV. For FESEM sample preparation aqueous solution of the test sample was put on a properly cleaned glass slide followed by spin coating. XRD of the sample was carried out in Bruker D8 instrument applying a primary beam monochromator to select the K α 1 component of the employed copper radiation (wavelength of 1.54056 Å). The samples for XRD were prepared by deposition of the samples on glass slide followed by drying in air. FTIR spectra were recorded in Perkin-Elmer spectrum 100. Prior to FTIR measurements the samples were ground with KBr (spectroscopic grade) and pressed to pellets. Surface morphology of mercury doped nano hydroxyapatite (nMHAP) was obtained by Atomic Force Microscpe NTEGRA (TS 150) in tapping mode. Hydrodynamic radius of the sample and particle size distribution (PSD) was measured by using Dynamic Light Scattering (DLS) in a particle size analyzer (90 Plus, Brookhaven Instrument Corporation, USA). Particle size was confirmed by using HRTEM of Phillips CM 200 at an operational voltage of 200 kV. For HRTEM micrograph a well dispersed aqueous solution of the sample was put into uniform copper coated TEM grid and well dried in air. For toxicological studies, OD of the samples was measured by using Shimadzu absorption spectrophotometer (Model No. 1700).

Cytotoxicity testing of MRC 5 and HepG2 cell lines by WST assay

This assay was based on enzymatic conversion of the tetrazolium salt WST-1 to formazan by cellular mitochondrial dehydrogenases present in viable cells. Briefly, cultured cells were diluted to 30,000 cells/ ml using an 8-channel pipette and then 100 μ l of well-mixed cell culture was added to each well except the last column on a 96-well plate. The plates were incubated in CO₂ incubator overnight. Then nMHAP was added at 25, 50, 100, 200 ppm concentration respectively in different well with positive (cell in media without any chemical treatment) and negative control (cell in media with EDTA treatment) by using 8-channel pipette dispenser. After

incubation for 72 hours, WST-1 proliferation reagent (Roche) was added to the cells (10 μ l per well) and incubation was continued for 4 hours at 37 °C. Plates were checked visually by comparing the colours of wells filled with treated cells (pink) and untreated cells showed orange colour. When a clear difference was seen by naked eye, results were checked by spectrophotometer at 490nm.

Dose response curves of in-vitro cytotoxicity data

Percentage of viable cells at different concentrations of nMHAP and in control wells gave different optical density value in spectrophometer. Dose response curves were plotted graphically from these plotted absorbance data.

Results and discussion

Particle size distributions (PSD) and hydrodynamic radius of synthesized mercury doped nano hydroxyapatite (nMHAP) was recorded from Dynamic Light Scattering by using a laser light (λ = 660 nm) scattered at an angle θ = 90^o at 25^o, placing the dispersion in a glass cuvette. A well diluted sample solution in milli-Q water was taken in a test tube and sonicated for 30 minutes in a sonication bath (Branson). After settling for 10 minutes the solution was again diluted with same ratio of milli-Q water and measurements were performed after sonication of the solution for 30 minutes. The average size was found to be 400 nm with poly dispersity index 0.255. DLS results were quite higher compared to actual size of the nanocomposite as it showed its hydrodynamic radius.

Figure 1a,b represented the low and high resolution FE-SEM image of the nMHAP. It revealed that the functionalized doped nanocomposite was completely rod shaped in nature, the inset image showed the individual rod like structure. The morphology was quite distinct however mixture of morphologies was not obtained in our micrograph.

Crystalline nature of the sample was determined by



XRD, prior to X-ray irradiation the samples were deposited on properly cleaned glass slides and dried well. Figure 2a demonstrated the XRD pattern of pure nano hydroxyapatite without any surface modification which showed 12 clearly visible peaks. The peaks obtained at $2\theta = 25.92^{\circ}$, 28.09° , 29.03°, 31.92°, 32.91°, 34.31°, 39.94°, 43.92°, 46.88°, 49.61°, 53.44°, 64.20° could be indexed to (002), (102), (210), (211), (300), (202), (130), (113), (222), (213), (004) and (323) diffraction planes of hydroxyapatite with a hexagonal structure [JCPDS card no. 74-0566] [26]. Whereas for nMHAP the peaks were almost unshifted with $2\theta = 25.92^{\circ}$, 28.09°, 29.03°, 31.92°, 32.85°, 34.13°, 39.57°, 43.77°, 46.69°, 49.43°, 53.25° and 64.01° indicating the same diffraction plane in figure 2b. Therefore the fabricated substrate possessed crystalline structure with doping of mercury ion in its amorphous form in to the crystalline structure of hydroxyapatite keeping the crystalline structure of native hydroxyapatite unchanged.

Successful surface functionalization on mercury doped hydroxyapatite was further confirmed by FTIR spectroscopy. FTIR spectra showed 7 characteristic peaks shown in figure 3, among which 3436 cm⁻¹ was for the O-H stretching. The peak obtained at 2914 cm⁻¹ signified C-H stretching and the C=O stretching frequency for carboxylic acid appeared at 1650 cm⁻¹. The peak appeared at 1380 cm⁻¹ and 1106 cm⁻¹ was for the characteristic of C-O-H bending and C-N stretching respectively. However the lower peak at 614 cm⁻¹ and 1036 cm⁻¹ was attributed to PO₄³⁻ and P-OH group respectively [26].

The surface morphology was obtained by atomic force microscopy which revealed the rod structure of the fabricated nanocomposite with average size 40 nm with average roughness of 9.44 nm. Figure 4a showed the AFM image and figure 4b showed its 3D diagram of surface morphology along with the histogram.

Appropriate particle size was determined by using high resolution transmission electron microscopy. Figure 5 showed the HRTEM micrograph which revealed that the fabricated Hg doped nanohydroxyapatite particles were rod shaped in nature. The average size of the particles ranged from 20 nm to 50 nm. The particles were distinctly visible and did not undergo agglomeration which also satisfied AFM results. Crystalline structure of the particle was further confirmed by SAED pattern (Figure 6). Chemical composition was confirmed by EDAX analysis, which showed that Ca, P, O and Hg were present as the major chemical component (**shown in supplementary information S1**) with nearly 23% of mercury present in the sample.

Exposure to nMHAP nanoparticle showed cytotoxic responses to MRC 5 cell line. In this case 76.44%, 71.87%, 58.39% and 34.98% cells were alive after 72 hours of incubation with 25, 50, 100 and 200 ppm concentration of

nMHAP respectively shown in figure 7a. In case of nMHAP treatment at these same dosages on HepG2 cell line, almost all cells were alive after 72 hours of the experiment, which means nMHAP is nearly nontoxic to HepG2 cells (Figure 7b). Though, mercury was itself toxic in nature, a specific carcinoma cell lines were affected by the nMHAP treatment. Moreover, doping with a bio-compatible constituent might have reduced its toxicity effectively. This unique result showed

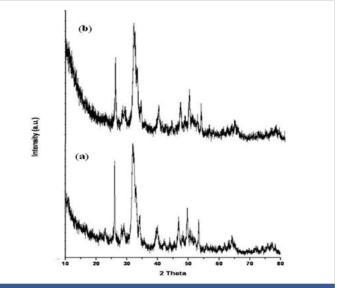


Figure 2: XRD pattern of (a) Hydroxyapatite, (b) Mercury doped hydroxyapatite.

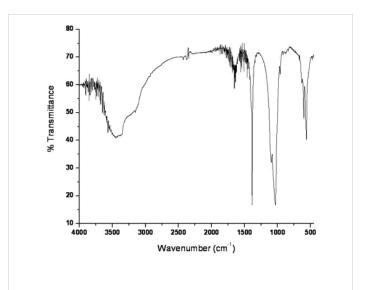


Figure 3: FTIR spectra of surface functionalized mercury doped hydroxyapatite.

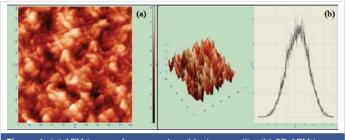


Figure 4: (a) AFM image of mercury doped hydroxyapatite, (b) 3D AFM image and histogram.



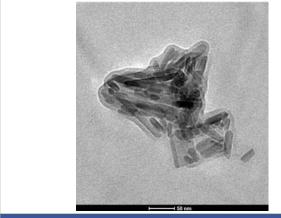
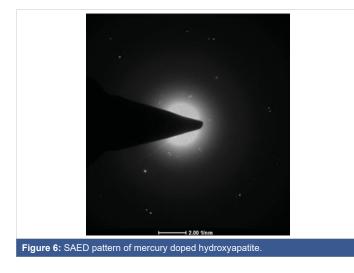
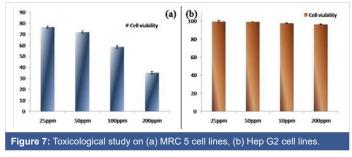


Figure 5: HRTEM image of nMHAP.

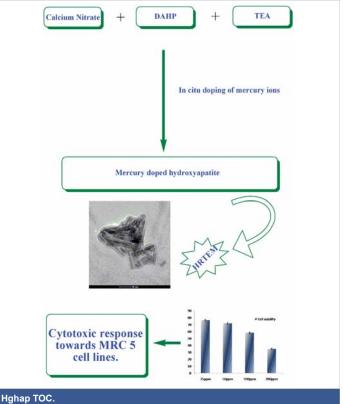




that cytotoxicity of a toxic component like mercury could be reduced to some extent by doping with a bio-compatible component. Cytotoxicity induced by nanoparticles has been attributed to a variety of mechanisms, which includes particle-induced apoptosis by up-regulation of Fas on the cell membrane, membrane damage and by intracellular effects associated with particle endocytosis. The upregulation of Fas indicates a link to anticancer activity [27]. This is certainly speculative at this point and needs further careful studies, especially to explain the selective toxicity to cancer cells.

Conclusion

This manuscript describes the doping of toxic mercury ions on to bio-compatible hydroxylapatite. Hydroxylapatite was synthesized by a novel sol-gel method comprising



mercury ion in situ. The nanocomposite was functionalized by grafting of carboxylic acid on to its surface by using PMIDA which was confirmed by FTIR study. High resolution TEM image showed that the particles were uniformly rod shaped with their size ranging from 20 nm to 50 nm. This nanoconposite showed unique site specific action on human fibroblast cell line MRC 5. Though nMHAP is toxic to MRC 5 cells, it showed no toxicity against liver carcinoma cell line HepG2. LD-50 for MRC 5 was found to be 133.13 ppm by regression method.

Though mercury was toxic in nature, it did not show toxicity against HepG2 cells. Liver is the organ for storage of harmful substances and removal of toxins. The popular chemotherapeutic drugs cause major damages to liver cells. As this nMHAP is nontoxic to hepatic cells, it may be the drug of choice for treating site specific carcinoma with minimum side effect.

Acknowledgement

We would like to thank ISI plan project for collaborative efforts and their generous financial support.

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